

Arsenic in the Environment: History, Sources and Perspectives

Kevin Armbrust

Department of Environmental Sciences

Louisiana State University

The classic poison.....



A bit of History.....

- Assassinations - White Arsenic (As_2O_3)
- Greeks and Romans
 - Lucius Corinilius Sulla (82 BC) – *Lex Cornelia*
- Renaissance and Middle Ages
 - Borgias and de Medici' families (14th 15th centuries)
 - Inheritance powder
 - Giulia Toffana/Hieronyma Spara (La Spara) (1600s) - Italy
 - Catherine Deshayes (La Voisin) –France
- Claire Booth Luce(U.S.) Ambassador to Italy (1950s)

So where is it?

- Lithosphere – 4×10^{13} tons
- Pedosphere – 2×10^9 tons
- Biosphere – 20,000 tons (mostly plants)
- Atmosphere – 800 – 1700 tons
- Hydrosphere
 - Rivers – 60,000 tons
 - Oceans – $2.4-5.2 \times 10^9$ tons

Approximate Fluxes

Input to Atmosphere	Input To Hydrosphere	Input to Pedosphere
Soil Bioproductivity – 26,000 t/a	Rivers to sea 54,000 – 90,000 t/a	Natural – 122,000 t/a
Volcanism – 17,000 t/a	Volcanism - 5000 t/a	Natural Weathering - ???
Seaspray – 27 t/a	Atmospheric Deposition – 5-8 t/a	Anthropogenic – 81,000 t/a
Forest fires – 3000 t/a	Soil weathering - ????	
Anthropogenic 25,000 t/a	Anthropogenic 125,000 t/a	

Duce, et al. 1991

Matschullat, 2000

Donet and Bruland, 1995

Atmosphere

85% N-hemisphere

15% S-hemisphere

680-1,480 t As
 0.5-2.8 ng m⁻³ (land)
 0.072-0.16 ng m⁻³ (ocean)

800-1,740 t As
 $\tau=0.022-0.027 \text{ a} \approx 7-10 \text{ days}$

120-260 t As
 1.0 ng m⁻³ (land)
 0.018 ng m⁻³ (ocean)

22,200-73,600 t a⁻¹

125 t a⁻¹ / 3,345 t a⁻¹ wood
 fuel and slash and burn nat.
 forest fire

4,300-8,200 t a⁻¹

27 t a⁻¹
 sea spray

Biosphere

Hydrosphere

25,450 t a⁻¹
 emissions

1.84 x 10⁵ t As terr. plants
 0.5 mg kg⁻¹

Anthropo-
 sphere

125,000 t a⁻¹ discharge
 28,400 t a⁻¹ residues

5.4-6.1 x 10⁴ t As
 3.6-61 x 10³ t a⁻¹
 dissolved
 720-12,240 t a⁻¹
 particulate
 0.1-1.7 μg L⁻¹

2.38-5.18 x 10⁹ t As
 2.33 x 10⁹ t As diss.
 19.9 x 10⁶ t As part.
 0.1-1.7 μg L⁻¹

1,980 t a⁻¹
 wind erosion

0.6-1.7 x 10⁹ t As
 5-7.5 mg kg⁻¹

?

160-26,200 t a⁻¹
 low T volatilization

Pedosphere

riverwater

oceans

4,870 t a⁻¹
 submarine
 volcanism

46,400 t a⁻¹ + 38,200 t a⁻¹
 sedimentation +
 subduction

17,150 t a⁻¹
 volcanoes

30,453 t a⁻¹
 As-production

??? t a⁻¹
 weathering

4.01 x 10¹³ t As in the earth crust

1.5-2 mg kg⁻¹ upper crust
 1-1.8 mg kg⁻¹ bulk crust

Lithosphere

imt 10/99

How are we exposed?

- **Air**
 - Rural areas (0.02-.2 ug/day)
 - Cities (0.4-0.6/day)
 - Cigarette smoke (0.8-2.4 ug/day)
- **Soil/Dirt** (0.14-0.28 ug/day)
- **Water** (1-5 ug/day)
- **Food**
 - Japan (126-273 ug/day)
 - US (60 ug/day)
 - Belgium (45 ug/day)
 - Canada (7 ug/day)*

How are we exposed?

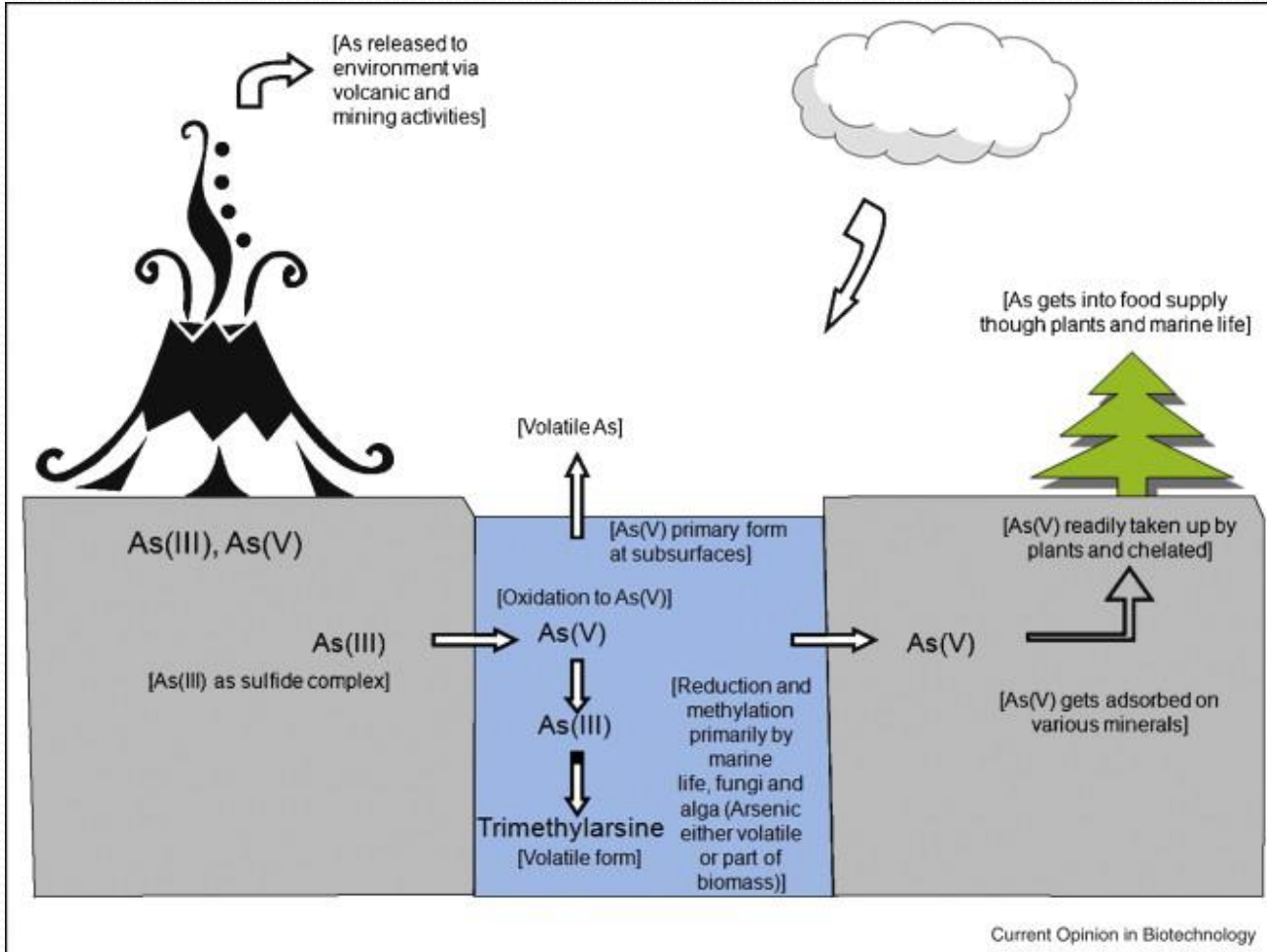
Food – (Most less than 0.25 mg/kg)

- **Seafood is highest**
 - Crustaceans (shrimp, crabs) – 30-100 mg/kg
 - Seaweed – 18-124 mg/kg
 - Fish – 2.4 – 17 mg/kg
 - Mollusks – 3- 20 mg/kg
- Mushrooms – 2.5 mg/kg
- Vegetables (misc) – 1.5 mg/kg
- Rice (various) – 0.2-0.9 mg/kg



Rose, et al. *Food Chem Tox.* 2007
J. Matschullat, *Sci. Total Env.* 2000

Sources in Soil



Transformations in Soil and Sediment

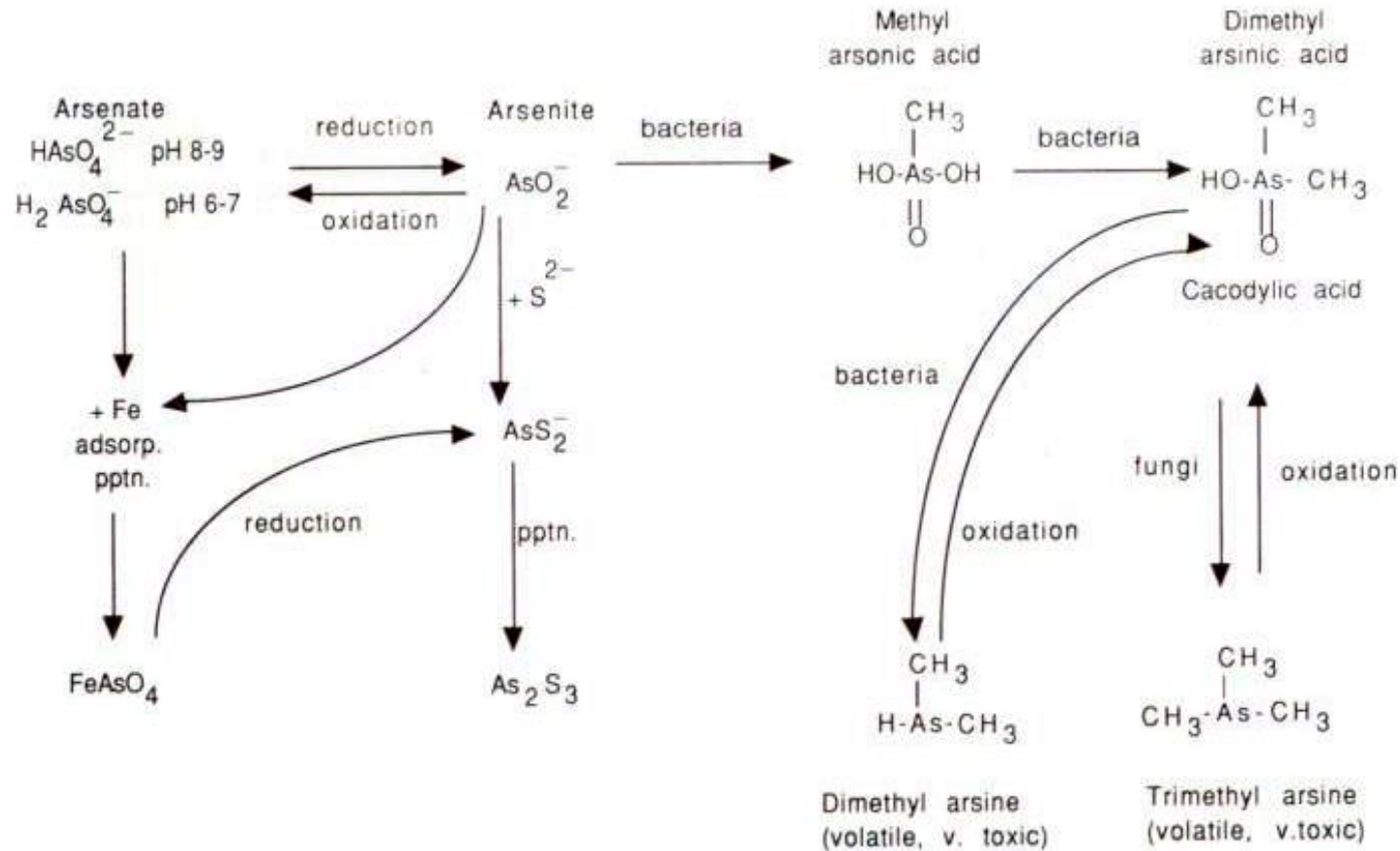


Figure 2. Chemical forms of arsenic and their transformations in soils.

Amounts in various plant species

- Radish – 30 mg/kg
- Spruce Bark – 10 mg/kg
- Mushrooms – 2.5 mg/kg
- Ferns – 1.3 mg/kg
- Clover – 0.330 mg/kg
- Lettuce – 0.25 mg/kg
- Rice and potatoes – 0.08 – 0.9 mg/kg
- Kale – 0.12 mg/kg
- Carrots – 0.08 mg/kg

Bergqvist, 2011 and Matschullat, 2000

As inputs to Soil (Anthropogenic)

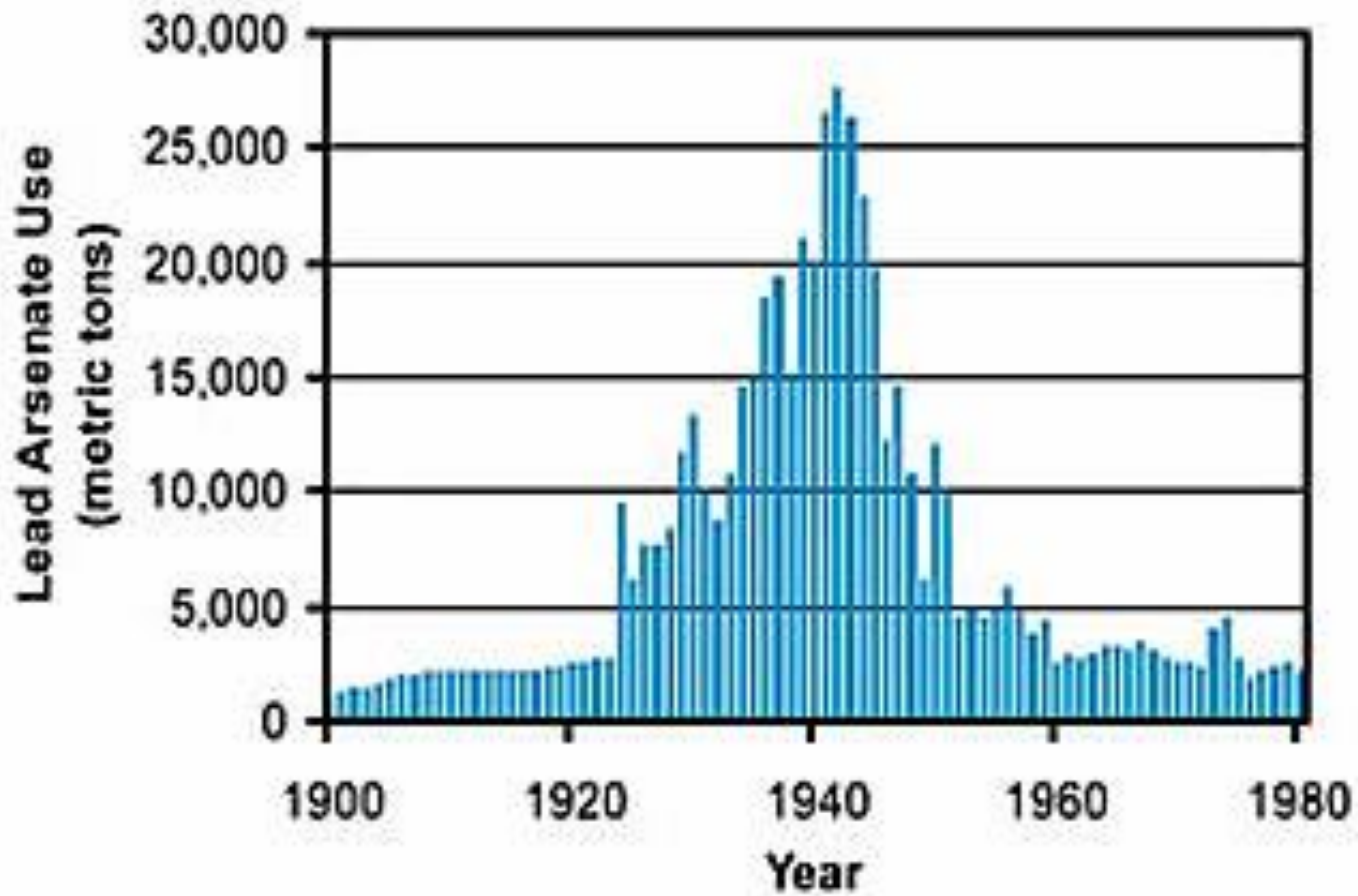
- Mining and smelting waste – 38,000 t/a
 - Water and sludge
- Coal ash – 22,000 t/a
- Atmospheric Deposition – 13,000 t/a
- Animal and Agricultural Waste – 5,800 t/a
 - Pesticides and fertilizers
- Logging and wood waste – 1,700 t/a

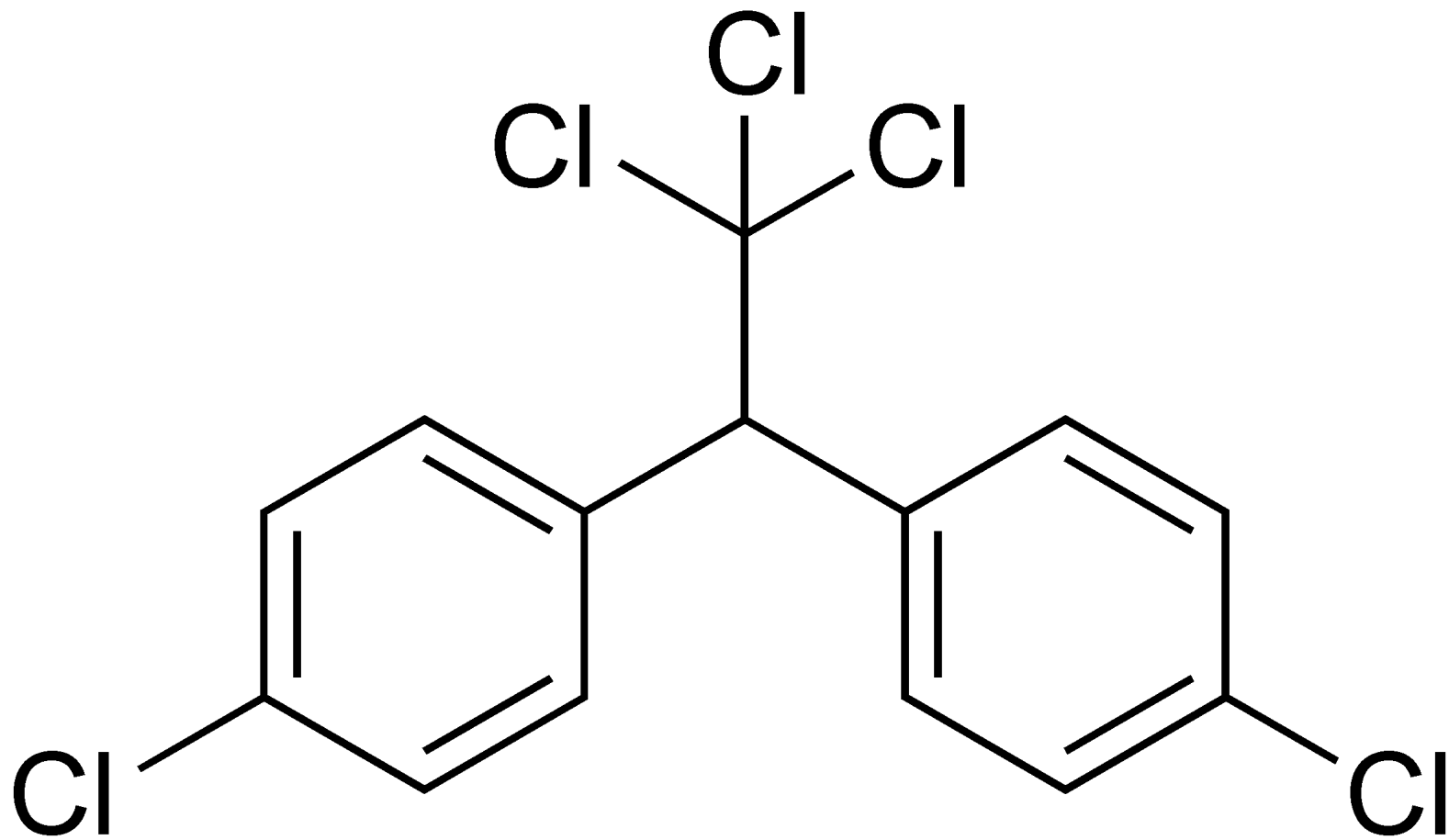
Arsenical pesticides

- Earliest Uses in 1800's (Paris Green)
 - Copper acetate triarsenite - Rodenticide

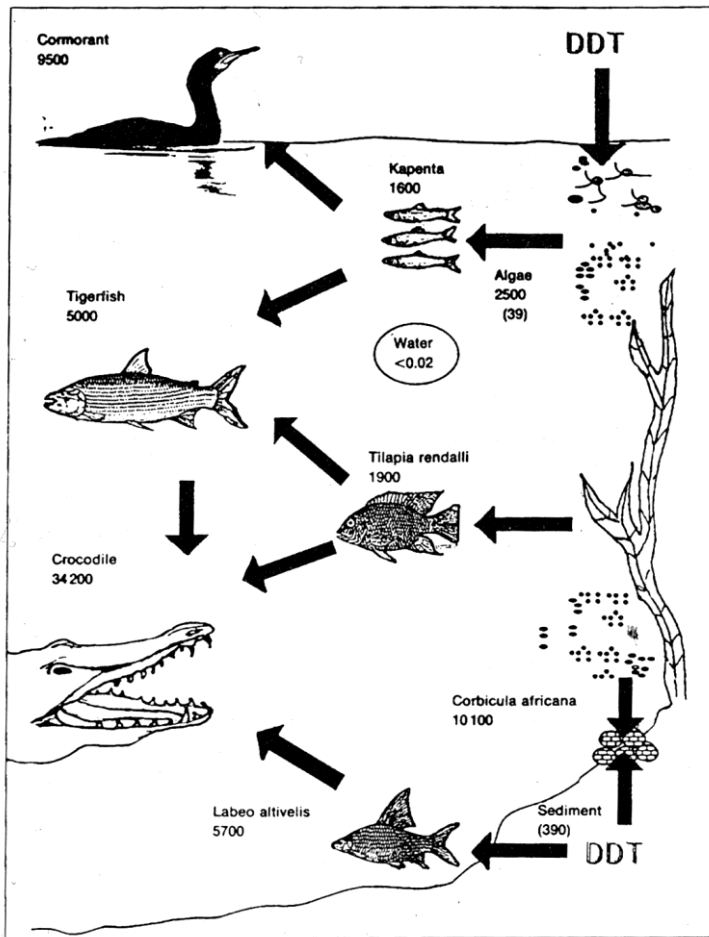












Arsenical pesticides

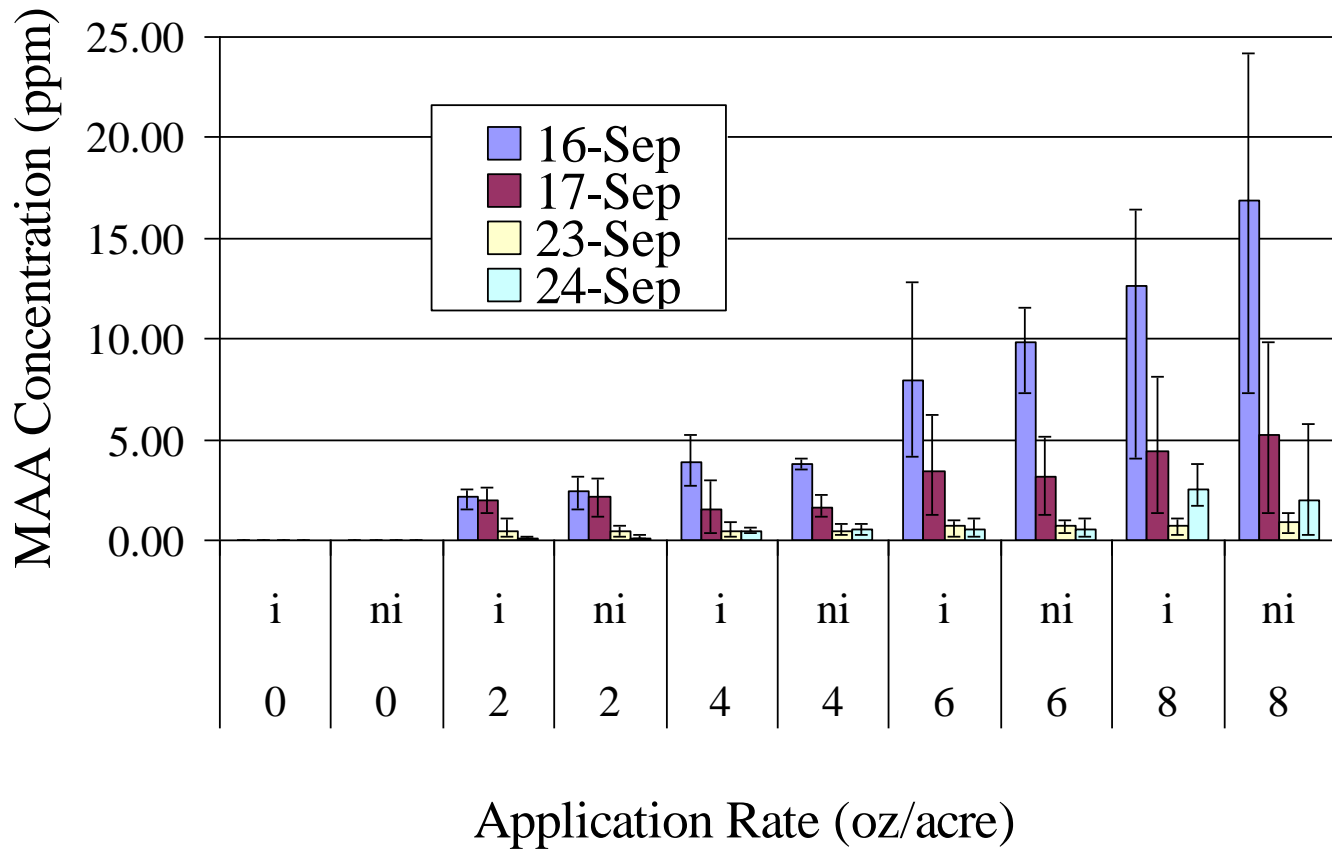
- Lead Arsenate - banned in 1998
- Chromated Copper Arsenate (CCA)
 - Wood preservative Banned in 2004
- MSMA, DSMA, CAMA, Cacodylic acid (DMAA)
 - Herbicides various AG, turf and ornamental and ROW.
- Roxarsone*

Dissipation of MSMA from Peanuts after Simulated Rainfall

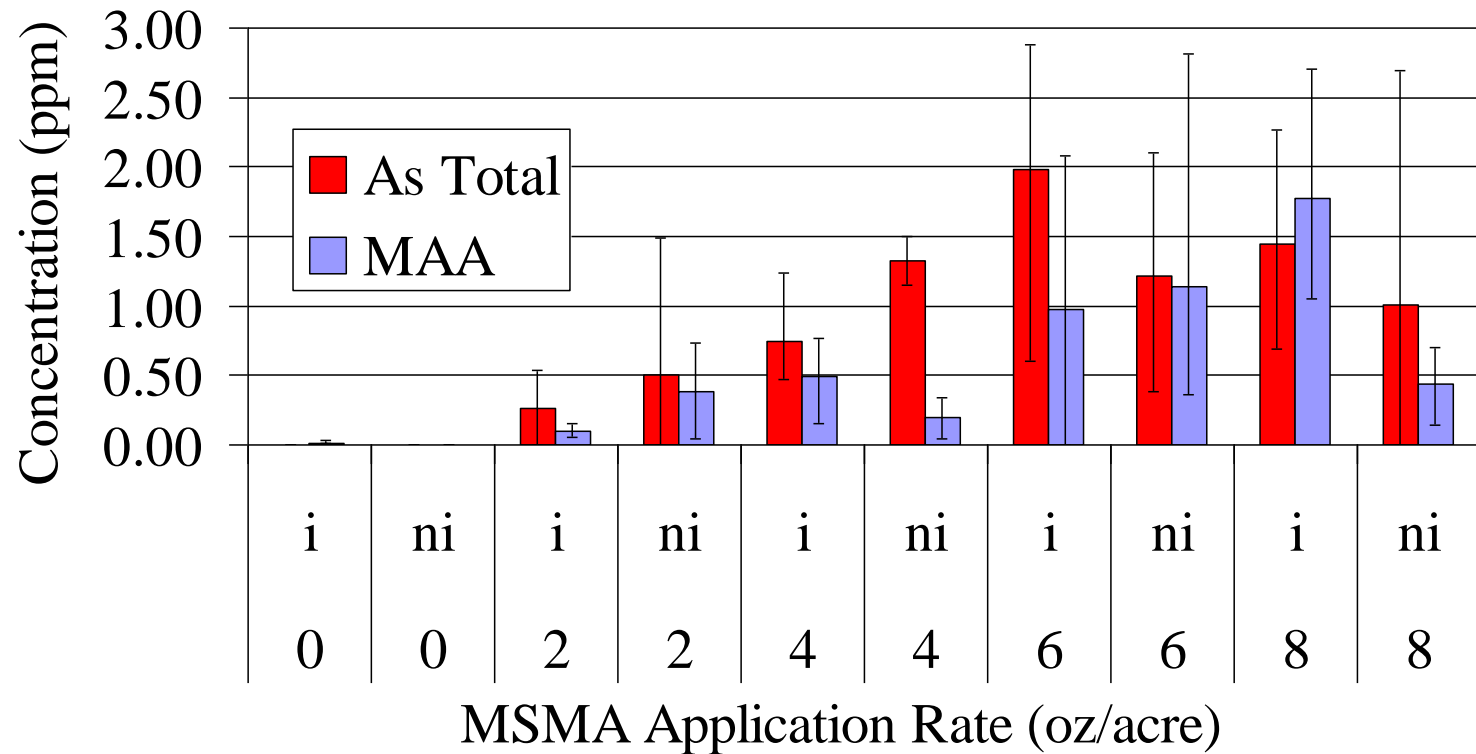
Kevin L. Armbrust and David Bridges
Department of Crop and Soil Sciences
University of Georgia – Griffin Campus
Georgia Department of Agriculture

Armbrust and Bridges, *J. Agric Food Chem.* 2002

Decline of Methyl Arsonic Acid (MAA) Residues on Peanut Leaves



MAA and Total Arsenic in Peanut Meat at Harvest



So where is arsenic?

Answer: Its everywhere..... Even in you!!!

- Kidneys – 0.005 – 1.5 mg/kg
- Liver – 0.02 – 1.6 mg/kg
- Muscle – 0.009 – 0.65 mg/kg
- Bones – 0.08 – 1.6 mg/kg
- Hair – 0.02 – 3.7 mg/kg
- Finger and toe nails – 0.2 – 3 mg/kg
- Blood and urine – 2 – 10 ug/L
- Overall average person – 0.14 – 0.26 mg/kg
 - Human reservoir on the planet – 108 tons

Summary and take home messages

- Arsenic is natural, exists in many forms in nature, and is mobilized by both man's activities and natural events.
- Greatest sources of inputs today due to mining and smelting activities, however agricultural activities have been historically important.
- Speciation of arsenic is critical. Differing species of As have different toxicities and different bioavailabilities. This will also impact any risk assessment.

Final thoughts.....

- Any risk assessment must include ALL aggregate sources of exposure for it to be of value.
- Route of exposure (air, water, food) and associated bioavailability is important.
- Consideration needs to be given to the analytical method, as certain extraction techniques can alter speciation.